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Introductory Chemistry: Concepts and Critical Thinking, 6th Edition © 2011 Pearson Education, Inc. Charles H. Corwin. EXAMPLE EXERCISE 3.2. Metric Unit Equations. Continued. Answers: (a) $1 \text{ Gm} = 1 \times 10^9 \text{ m}$ (b) $1 \text{ g} = 1 \times 10^{-2} \text{ cg}$ (c) $1 \text{ L} = 1 \times 10^6 \text{ }\mu\text{L}$ (d) $1 \text{ s} = 1 \times 10^3 \text{ ms}$. Complete the unit equation for each of the following exact metric equivalents:

EXAMPLE EXERCISE 3.1 Metric Basic Units and Prefixes

Although the sixth edition incorporates many impressive features, such as macro to micro artwork, animations correlated to the text, and hand-sketched worked examples, General Chemistry is still 200 to 300 pages shorter and much less expensive than other two-semester textbooks. Dr.

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