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2 (g) + 2H 2 O ( ) CH 4 (g) + 2O 2 (g) During the reaction the temperature of the reaction mixture decreases. 6.1 Define the term enthalpy change. (2) 6.2 Does the enthalpy change ( H) for this reaction have a positive or negative value? Explain the answer by referring to the energy involved. (2)